

## Instrumental Effects in Secondary Electron Yield and Energy Distribution Measurements<sup>†\*</sup>

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<sup>†</sup>Work supported by the U.S. Department of Energy under contract number DE-AC03-76SF00515 (SLAC).

\*Presented at the **31st ICFA Advanced Beam Dynamics Workshop on Electron-Cloud Effects**, April 19-23, 2004, Napa, Ca.

5/11/2004



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## Effects

- Secondary "primary" electrons generated inside the source
- Secondary electrons generated inside RFA analyzer or from the chamber
- Surface modification by incident electrons (desorption, carburization, oxidation, damage)
- Substrate effects
- Near-zero energy



#### **Energy Distribution (ED) of Secondary Electrons**





## Secondary Electron Generation





#### **Extruded-AI Beam Chamber Topography**





#### **SEY Measurement - RFA**



Strengths: Angular and energy distribution measurements possible Weaknesses: Grid/collimator tertiaries; gun space charge



#### **SEY Measurement - Sample Current**



Strengths: Angular measurements; no stray secondaries (with -20V) Weaknesses: Yield does not include elastics; gun space charge; tertiaries from surrounding chamber



#### **SEY Measurement - Sample Retard**



Strengths: Simple equipment (no space charge limit with gun);tertiaries rejected after -20V

Weaknesses: No angular measurements; yield does not include elastics



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### Secondary "Primary" Electrons





#### Secondary "Primary" Electrons





## **Beam Current Profile**

#### FC Aperture = 0.25 mm





### Secondary "Primary" Electrons





#### **Secondary "Primary" Electrons**





#### **Unipotential Electron Gun**



Simple electronics, but space charge problem below 200 eV



### **Fixed Element Voltages**



Good performance to < 10 eV, expensive, complex design



### Secondary "Primary" Electrons





Yes, but from the 0 V shield!





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### "Environmental" Tertiaries





### **RFA Tertiary Electrons**





## Effects

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- Tertiary electrons generated inside RFA analyzer or from the chamber
- Surface modification by incident electrons (desorption, carburization, oxidation, damage)
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#### **Causes of Electron-Induced SEY Reduction**

- Electron desorption of surface gases, particularly importantly barrier-reducing water and hydrocarbons
- Dissociation of aromatic HCs to low-yield polymerized carbon
- Electron-reduction of high-yield oxides
- Electron-activated grain boundary diffusion of carbon in the presence of hydrogen



#### **Carbon Grain Boundary Diffusion**

- Observed on aluminum covered with native oxide and thin γ-alumina
- Surface carbon was produced from electron reduction of CO, both from gas phase and from AI bulk, up the grain boundaries
- Co-adsorption of H<sub>2</sub> increased the surface concentration of CO at hydroxyl sites

From Garwin et al, SLAC Pubs. 392 (1968) and 2716 (1981).



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## Primary Electron Range (TiN)

(All axes in angstroms)



$$\theta = 0^{\circ}$$
  $\Theta = 83^{\circ}$   $\Theta = 83^{\circ}$ 

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### Substrate Effect





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#### Yield From Sputtered (But Disordered) Surfaces





# **Elastic Reflection**

